Kinetic Monte Carlo Simulation of Electron Transfer

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Biological Electron Transfer

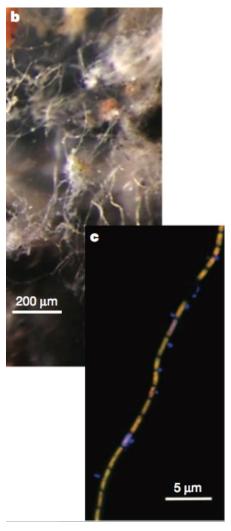
Filamentous bacteria transport electrons over centimetre distances

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Oxygen consumption in marine sediments is often coupled to the oxidation of sulphide generated by degradation of organic matter in deeper, oxygen-free layers. Geochemical observations have shown that this coupling can be mediated by electric currents carried by unidentified electron transporters across centimetre-wide zones. Here we present evidence that the native conductors are long, filamentous bacteria. They abounded in sediment zones with electric currents and along their length they contained strings with distinct properties in accordance with a function as electron transporters. Living, electrical cables add a new dimension to the understanding of interactions in nature and may find use in technology development.



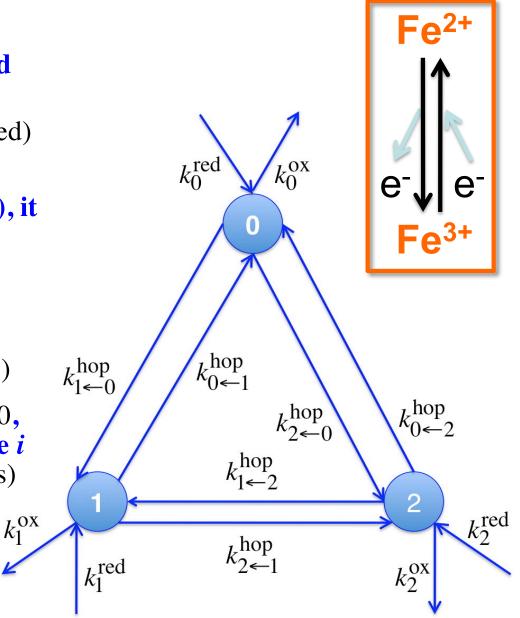


Model Definition

- A set of *N* redox molecules
- Each molecule $i \in [0, N-1]$ can hold up to one electron:

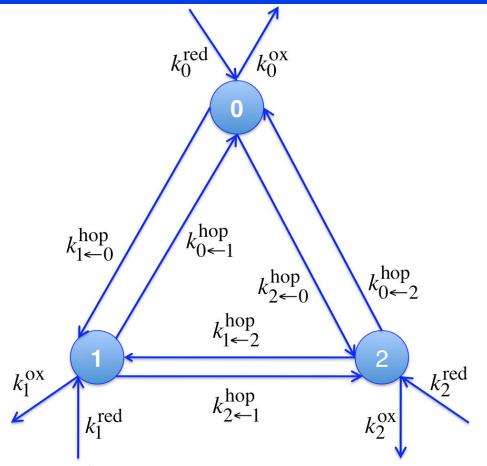
occ(i) = 1 (occupied) or 0 (unoccupied)

- Reduction: If molecule i is not occupied by an electron (occ(i) = 0), it can be reduced (or an electron is injected to it) at a rate of k_i^{red} (1/s)
- Oxidation: If occ(i) = 1, molecule i can be oxidized (or the electron is ejected from it) at a rate of k_i^{ox} (1/s)
- *Hopping*: If occ(i) = 1 and occ(j) = 0, the electron can hop from molecule i to molecule j at a rate of $k_{i\leftarrow i}^{\text{hop}}$ (1/s)



Molecular Network Topology

- Directed graph consisting of N nodes (= molecules) and directed edges (= possible hopping from a node to a neighbor node)
- ngb(i): Number of outgoing edges for molecule i (Max_ngb = maximum number of outgoing edges per node)
- $lsngb[N][Max_ngb+1]$ lsngb[i][0] = ngb(i) lsngb[i][k] = molecular ID of the k-thoutgoing neighbor of molecule i, where $k \in [1, ngb(i)]$



Example: $Max_ngb = 2$

lsngb[i][k]	k = 0	1	2
i = 0	2	1	2
1	2	0	2
2	2	0	1

Data Structures

Occupation

$$occ[N]: occ[i] = \begin{cases} 1 & \text{molecule } i \text{ is occupied by an electron} \\ 0 & \text{unoccupied} \end{cases}$$

Rates (constants)

$$rate[N][Max _ngb + 2]$$

$$\begin{cases} rate[i][0] = k_i^{\text{red}} \\ rate[i][1] = k_i^{\text{OX}} \\ rate[i][k+1] = k_{lsngb[i][k] \leftarrow i}^{\text{hop}} \end{cases}$$

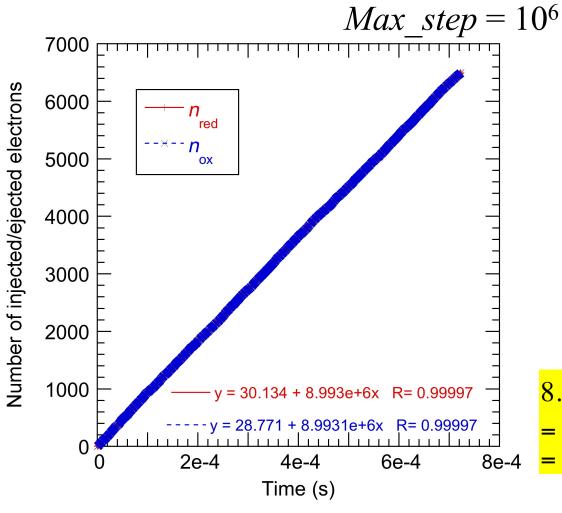
Occupation-modified rates (dynamic variables)

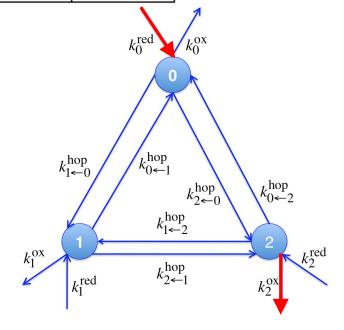
Algorithm

```
occ[i] \leftarrow 0 \ (i = 0 \text{ to } N-1)
n_{\rm red} \leftarrow 0 // number of injected electrons
n_{\rm ox} \leftarrow 0 // number of ejected electrons
t \leftarrow 0
for step \leftarrow 1 to Max step // Max step = total number of KMC steps
   r \leftarrow 0
   for i \leftarrow 0 to N-1
      r += (rate \ occ[i][0] = rate[i][0]*(1-occ[i]))
      r += (rate\_occ[i][1] = rate[i][1]*occ[i])
                                                                                               r = \sum_{i=1}^{n} r_{event}
      for k \leftarrow 1 to lsngb[i][0]
          r += (rate \ occ[i][k+1] = rate[i][k+1]*(1-occ[lsngb[i][k]])*occ[i])
   t = \ln(\text{rand}()/RAND \ MAX)/r
                                                                                       \Delta t = -\ln(u_{\rm random})/r
   r_{\rm th} \leftarrow r^* {\rm rand}()/RAND \ MAX
   r_{\rm acc} \leftarrow 0
   for i \leftarrow 0 to N-1
      if (r_{th} < (r_{acc} += rate\_occ[i][0]) // reduction occurs
                                                                                                        r_{\text{th}}
           occ[i] \leftarrow 1; print t, ++n_{red}, n_{ox}; break
      else if (r_{th} < (r_{acc} += rate\_occ[i][1]) // oxidation occurs)
                                                                                                      r_2
                                                                                            r_1
           occ[i] \leftarrow 0; print t, n_{red}, ++n_{ox}; break
       else
          for k \leftarrow 1 to lsngb[i][0]
             if (r_{th} < (r_{acc} += rate \ occ[i][k+1]) // \text{ hopping from } i \text{ to } k\text{-th neighbor occurs})
                           occ[lsngb[i][k]] \leftarrow 1; occ[i] \leftarrow 0; break
      if (r_{th} < r_{acc}) break
```

Example Run

rate[i][k] (1/s)	k = 0	1	2	3
i = 0	1e7	0	1e9	2e9
1	0	0	3e9	1e7
2	0	2e7	2e7	4e9





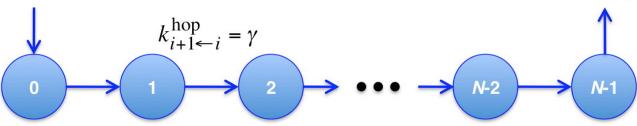
Electric current

$$8.99 \times 10^{6} (1/s) \times 1.602 \times 10^{-19} (C)$$

= $1.44 \times 10^{-12} (C/s)$
= $1.44 (pA)$

Asymmetric Simple Exclusion Process

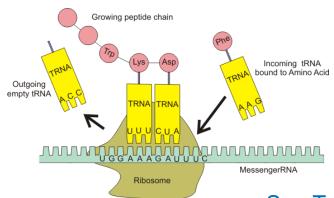
- Linear chain of N molecules
- *Reduction* can occur only at molecule 0 with $k^{red} = \alpha$ (1/s)
- Oxidation can occur only at molecule N-1 with $k^{ox} = \beta$ (1/s)
- Hopping can occur from molecule *i* to *i*+1 ($i \in [0, N-2]$) with $k_{i+1\leftarrow i}^{\text{hop}} = \gamma$ (1/s) $k^{\text{red}} = \alpha$



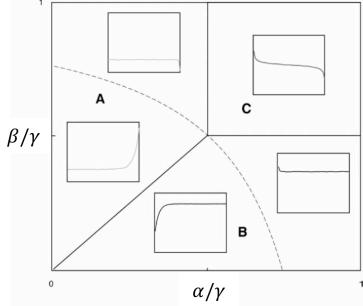
ASEP:

"Ising model of nonequilibrium statistical physics"

- Nonequilibrium phase transition from low-density (LD) to high-density (HD) phase with increasing α/β
- Applications in traffic flow and ribosome motion on mRNA during translation



Peptide Synthesis



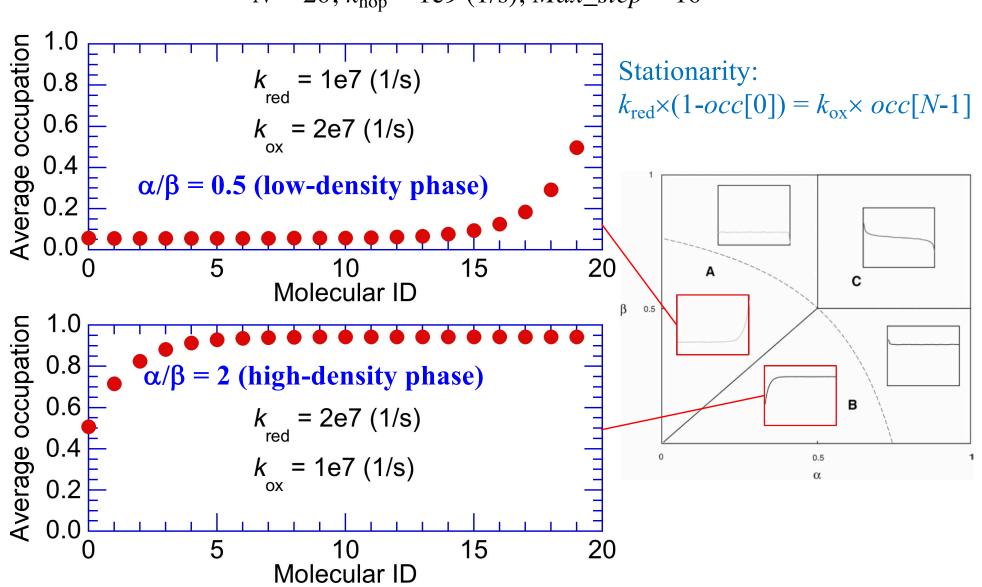
A. Shadschneider, Physica A 285, 101 ('00)

See <u>Taming nonequilibrium statistics</u> & <u>cytoskeletal traffic</u>

ASEP Occupation Profile

KMC simulation results

$$N = 20$$
; $k_{\text{hop}} = 1e9 (1/s)$; $Max_step = 10^6$

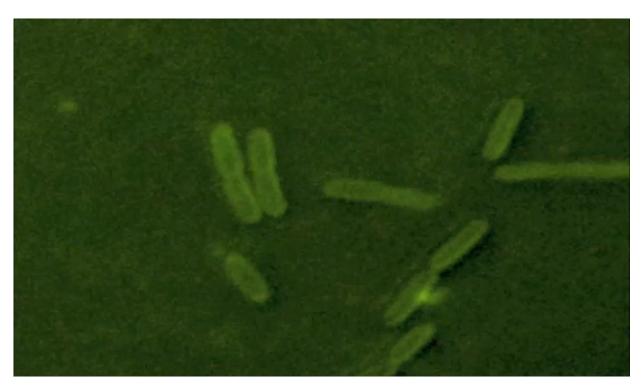


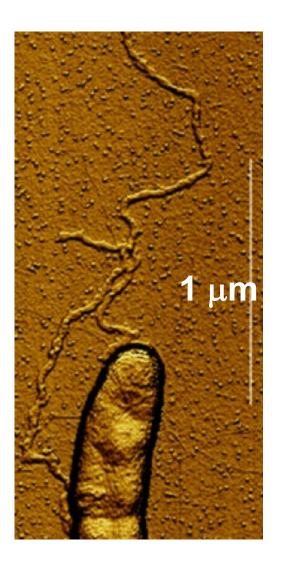
Algorithm for KMC Simulation of ASEP

```
k^{\text{red}} = \alpha \quad (k^{\text{red}}, k^{\text{ox}}) = (1e7 \text{ s}^{-1}, 2e7 \text{ s}^{-1}) \text{ or } (2e7 \text{ s}^{-1}, 1e7 \text{ s}^{-1}) \quad k^{\text{ox}} = \beta
occ[i] \leftarrow 0 \ (i = 0 \text{ to } N-1) \quad N = 20
                                                                                k_{i+1 \leftarrow i}^{\text{hop}} = \gamma \quad \text{1e9 (s-1)}
n_{\rm red} \leftarrow 0 // number of injected electrons
n_{\rm ox} \leftarrow 0 // number of ejected electrons
t \leftarrow 0
                   Max step = 10^6
for step \leftarrow 1 to Max step // Max step = total number of KMC steps
   r \leftarrow 0 // Compute total escape rate
   for i \leftarrow 0 to N-1
       r += (rate \ occ[i][0] = rate[i][0]*(1-occ[i]))
                                                                                                         r = \sum_{i=1}^{n} r_{event}
       r += (rate \ occ[i][1] = rate[i][1]*occ[i])
       for k \leftarrow 1 to lsngb[i][0]
                                                                                                               event
          r += (rate\_occ[i][k+1] = rate[i][k+1]*(1-occ[lsngb[i][k]])*occ[i])
   t = \ln(\text{rand}()/RAND \ MAX)/r \ // Advance time
                                                                                                \Delta t = -\ln(u_{\rm random})/r
   r_{\rm th} \leftarrow r^* {\rm rand}()/RAND~MAX~// Pick the next state
   r_{\rm acc} \leftarrow 0
   for i \leftarrow 0 to N-1
       if (r_{th} < (r_{acc} += rate\_occ[i][0]) // reduction occurs
                                                                                                               I_{\mathsf{th}}
           occ[i] \leftarrow 1; print t, ++n_{red}, n_{ox}; break
       else if (r_{th} < (r_{acc} += rate\_occ[i][1]) // oxidation occurs
           occ[i] \leftarrow 0; print t, n_{red}, ++n_{ox}; break
       else
          for k \leftarrow 1 to lsngb[i][0]
              if (r_{th} < (r_{acc} += rate \ occ[i][k+1]) // hopping from i to k-th neighbor occurs
                             occ[lsngb[i][k]] \leftarrow 1; occ[i] \leftarrow 0; break
       if (r_{\rm th} < r_{\rm acc}) break
```

Application: Bacterial Nanowires

- Electron transfer (ET) governs all known energy-conversion (redox) processes in biology
- A remarkable example is the discovery of rapid ET along bacterial nanowires produced by *Shewanella oneidensis* MR-1

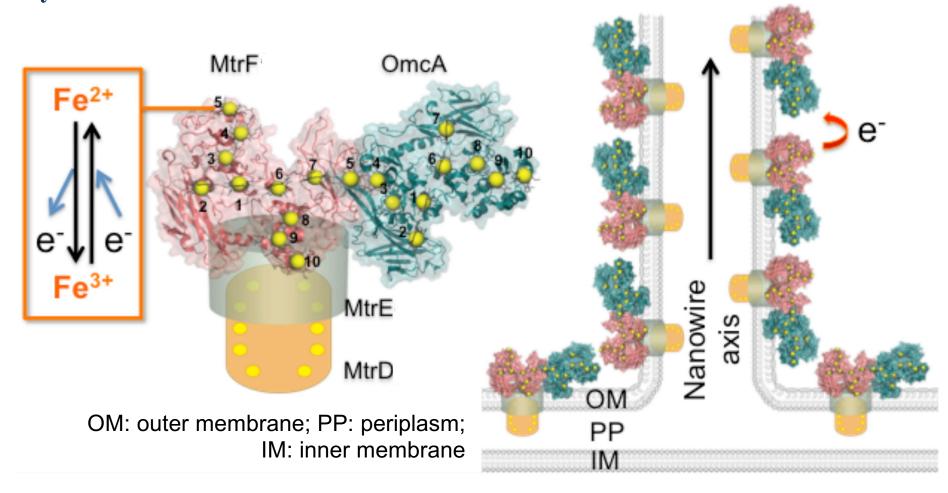




Shewanella emitting electrically conducting nanowires

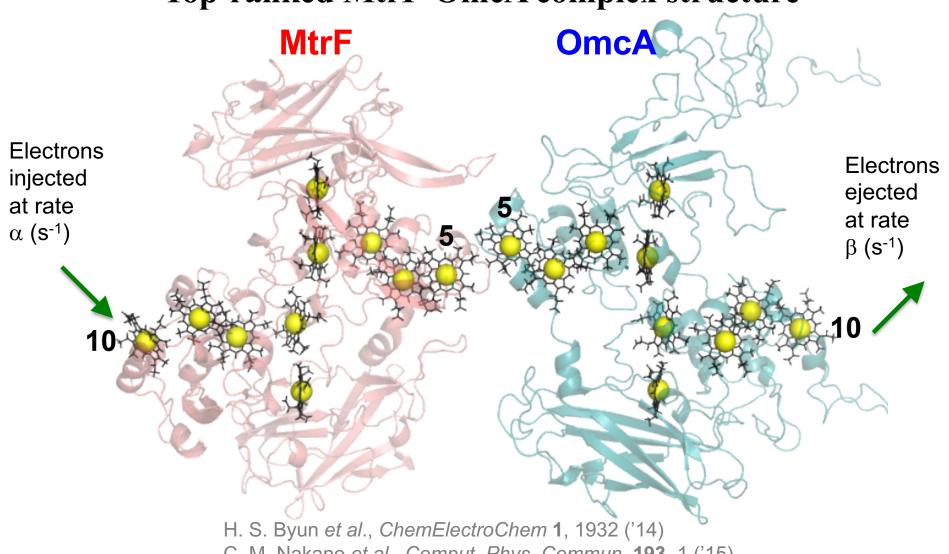
Scientific Problem & Research Goal

- Outer-membrane cytochromes, MtrF & OmcA, are hypothesized media for ET, but how they are assembled into a conducting complex remains a mystery
- Goal: Determine the structure of MtrF-OmcA complex & visualize ET dynamics in it to understand electric conduction mechanisms



Result: Complex Structure

Top-ranked MtrF-OmcA complex structure



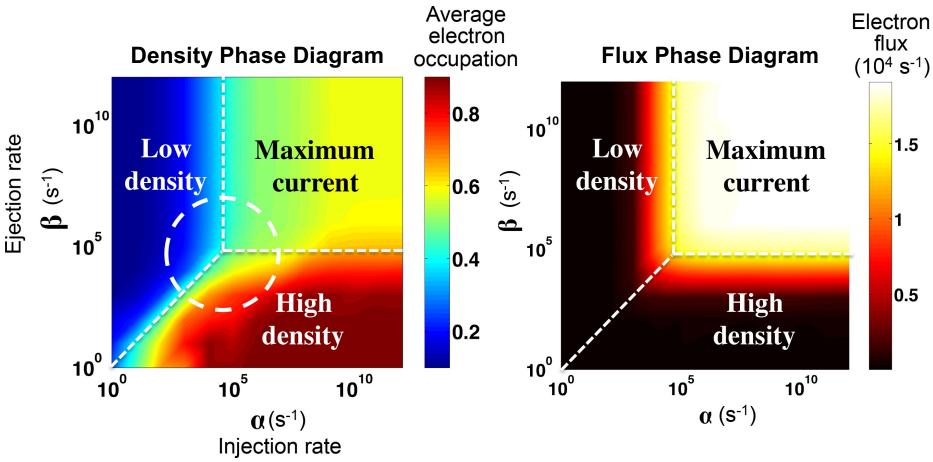
C. M. Nakano et al., Comput. Phys. Commun. 193, 1 ('15)

C. M. Nakano et al., J. Mol. Graph. Model. 65, 94 ('16)

T. Wei et al., J. Phys. Chem. Lett. 7, 929 ('16)

H. Byun et al., Comput. Phys. Commun. 219, 246 ('17)

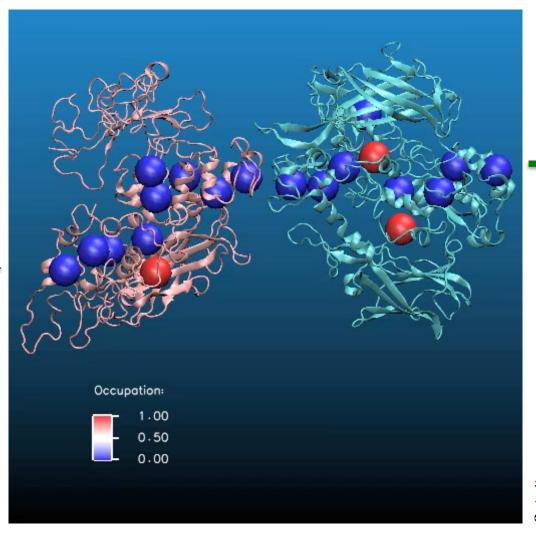
Result: ET Phase Diagram



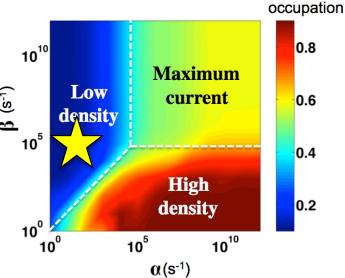
- KMC results exhibit a nonequilibrium phase transition from low-density to high-density phases for increased ratio of electron injection rate (α) to ejection rate (β)
- When both α & β exceed the smallest k_{ij} , another transition to the maximum-current phase was found
- Observed respiration rates (10³-10⁴ s⁻¹) indicate that "life operates around the triple phase junction," thus a small change in the electrochemical environment effectively triggers a large electric response

Animation: Low-Density Phase

Electron injection rate: $\alpha = 10^2 \text{ s}^{-1}$



Electron ejection rate: $\beta = 10^5 \text{ s}^{-1}$

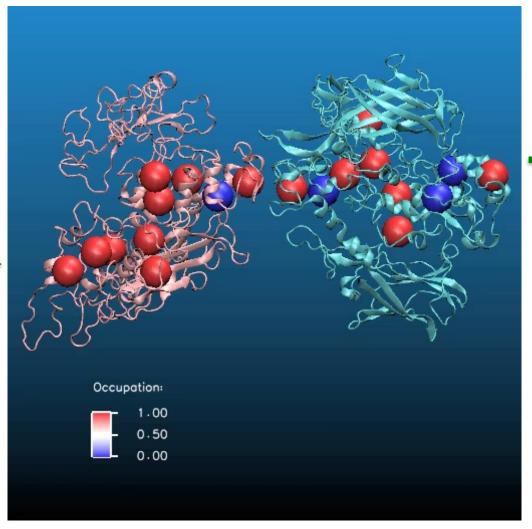


Average electron

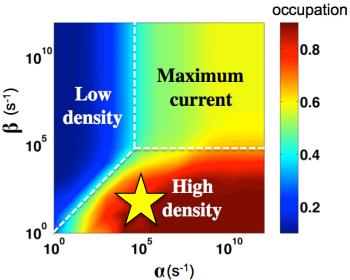
• Electron flow is limited by the small number of electrons

Animation: High-Density Phase

Electron injection rate: $\alpha = 10^5 \text{ s}^{-1}$



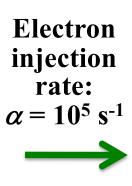
Electron ejection rate: $\beta = 10^2 \text{ s}^{-1}$

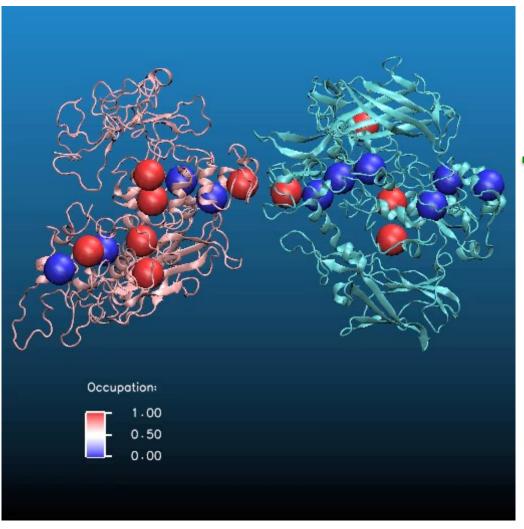


Average electron

• Electron flow is limited by the congestion of electrons

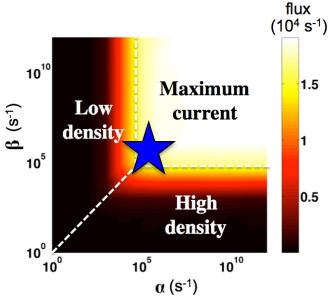
Animation: Maximum-Current Phase





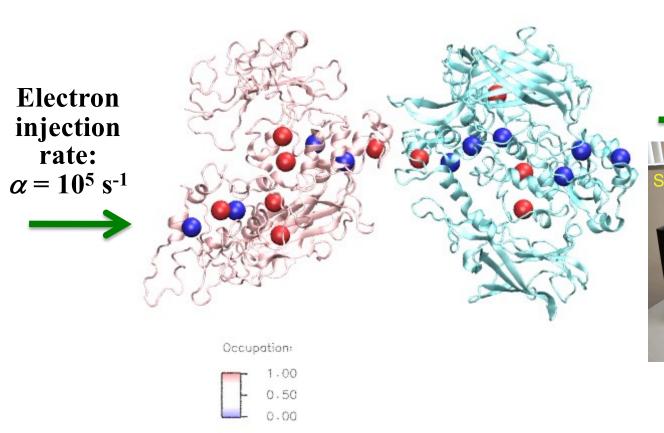
Electron ejection rate: β = 10⁵ s⁻¹

- Large electric current is facilitated by balanced electron injections & ejections
- Life operates around the triple junction

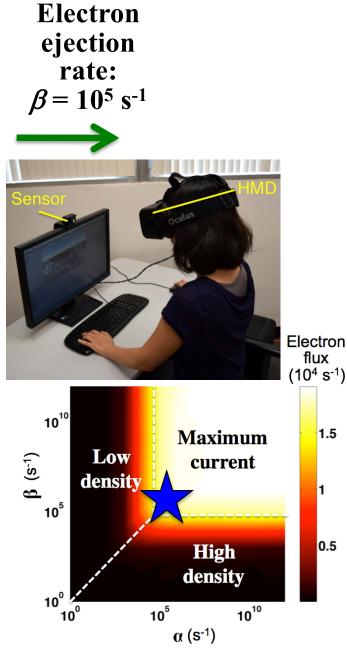


Electron

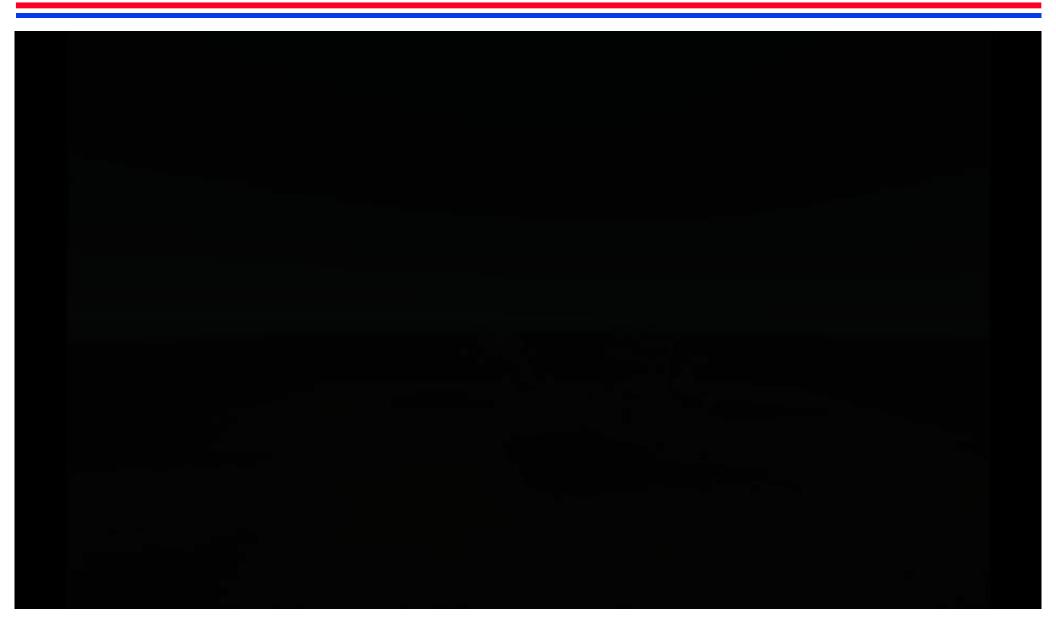
Animation: Maximum-Current Phase



• Each electron-transfer event is represented by a directed edge



Navigation in Virtual Reality



C. M. Nakano et al., J. Mol. Graph. Model. 65, 94 ('16)

GEARS

GEARS (Game-engine-assisted research platform for scientific computing) allows users to develop & perform immersive & interactive simulations within commodity virtual reality (VR) platforms



Oculus Rift + Leap Motion



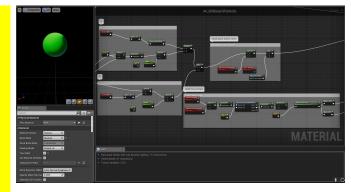
HTC Vive



Photoexcitation in MoSe₂

Accomplishments:

- Implemented simulation workflows in VR-capable Unity & Unreal game engines
- Enhanced interaction utilities, e.g., virtual confocal microscopy
- Developed an interface with community MD software, LAMMPS, & demonstrated immersive & interactive 250K-atom simulations on desktop

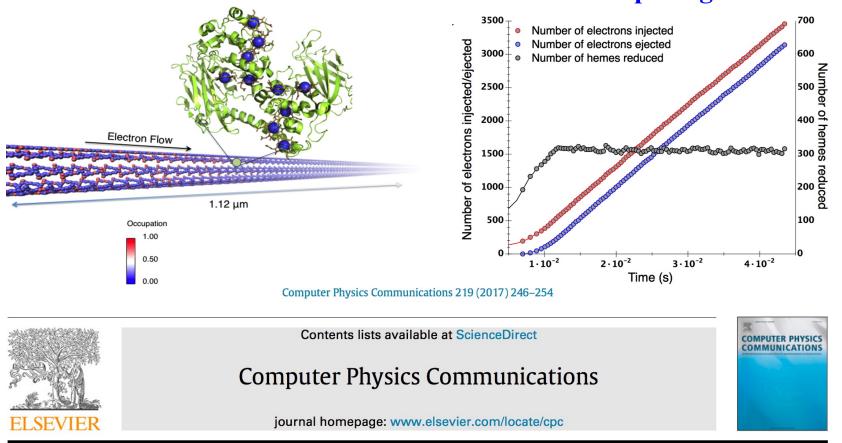


LammpsVR editor



Scalable Parallel KMC

 93.5% of perfect speedup for a 4.2 billion-heme system on 1,024 Intel Xeon processors at USC Center for Advanced Research Computing



A derivation and scalable implementation of the synchronous parallel kinetic Monte Carlo method for simulating long-time dynamics



Hye Suk Byun ^a, Mohamed Y. El-Naggar ^{a,b,c}, Rajiv K. Kalia ^{a,d,e,f}, Aiichiro Nakano ^{a,b,d,e,f,*}, Priya Vashishta ^{a,d,e,f}