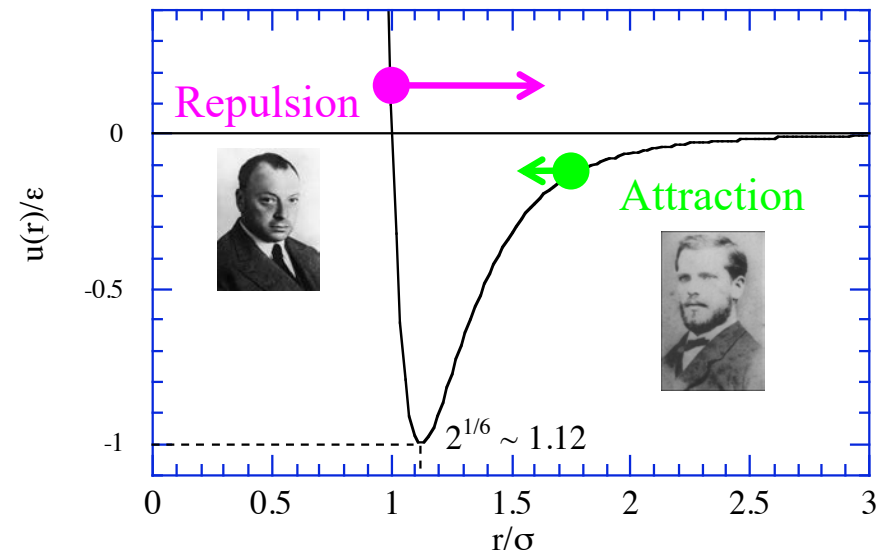


Recap: Lennard-Jones Potential

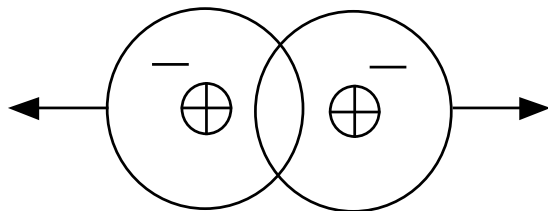
$$V(\vec{r}^N) = \sum_{i < j} u(r_{ij}) = \sum_{i=0}^{N-2} \sum_{j=i+1}^{N-1} u(|\vec{r}_{ij}|)$$

where $\vec{r}_{ij} = \vec{r}_i - \vec{r}_j$ and

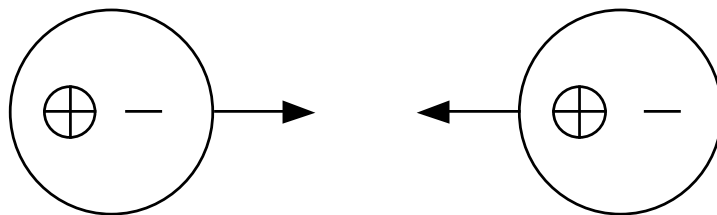
$$u(r) = 4\epsilon \left[\left(\frac{\sigma}{r} \right)^{12} - \left(\frac{\sigma}{r} \right)^6 \right]$$



Short-range repulsion by Pauli exclusion between electrons



Long-range attraction by polarization



Johannes D. van der Waals
Nobel Physics Prize (1910)



Wolfgang Pauli
Nobel Physics
Prize (1945)

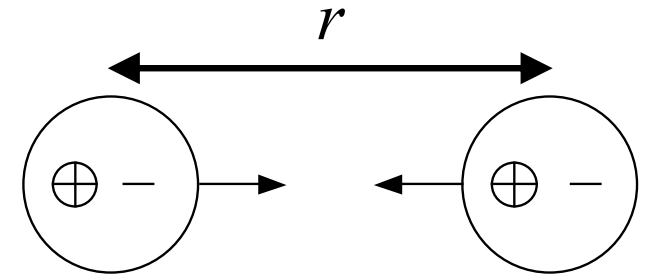


John E. Lennard-Jones
PPS 43, 461 (1931)

Q: Where Are the 12-6 Powers from?

$$u(r) = 4\varepsilon \left[\left(\frac{\sigma}{r} \right)^{12} - \left(\frac{\sigma}{r} \right)^6 \right]$$

Long-range attraction by polarization: electrostatics



Multipole expansion of electrostatic potential (*cf.* fast multipole method lecture)

<https://aiichironakano.github.io/phys516/FMM.pdf>

Also, Legendre generating function

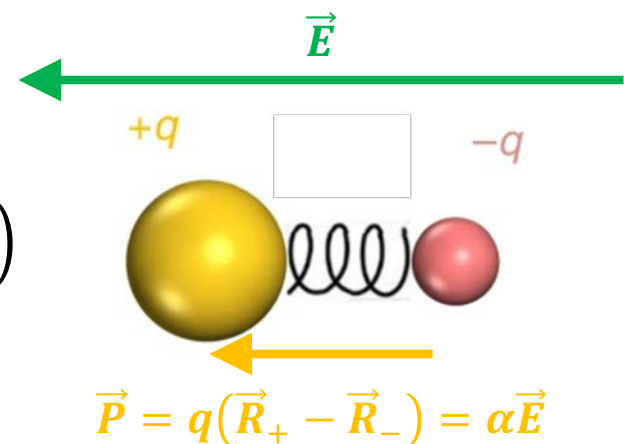
$$\frac{1}{|\vec{r} - \vec{r}'|} = \frac{1}{r} \sum_{j=0}^{\infty} P_j(\cos\theta) \left(\frac{r'}{r} \right)^j$$

Electrostatic potential: $\phi(r) \sim \underbrace{O\left(\frac{1}{r}\right)}_{\substack{\oplus + \ominus = 0 \\ \propto \text{charge}}} + \underbrace{O\left(\frac{1}{r^2}\right)}_{\substack{\oplus \leftarrow \ominus = \vec{P} \\ \propto \text{dipole}}} + \dots = O\left(\frac{1}{r^2}\right)$ for neutral atom

Electric field: $\vec{E}(\vec{r}) = -\frac{\partial \phi}{\partial \vec{r}} = O\left(\frac{1}{r^3}\right)$

Induced dipole: $\vec{P}(\vec{r}) = \underbrace{\tilde{\alpha}}_{\text{polarizability}} \vec{E}(\vec{r}) = O\left(\frac{1}{r^3}\right)$

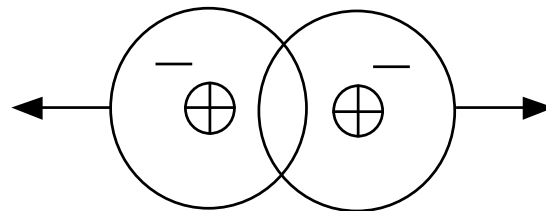
Dipole energy: $-P(\vec{r}) \cdot E(\vec{r}) \sim \frac{1}{r^3} \cdot \frac{1}{r^3} = O\left(\frac{1}{r^6}\right)$



Q: Where Are the 12-6 Powers from?

$$u(r) = 4\varepsilon \left[\left(\frac{\sigma}{r} \right)^{12} - \left(\frac{\sigma}{r} \right)^6 \right]$$

Short-range repulsion by Pauli exclusion between electrons



To these, London's dispersive (attractive) forces are added: these are instantaneous dipole-induced dipole interactions between non-polar molecules. They act at relatively short distances, distinctly longer, however, than the repulsive forces.

There is no theoretical equation describing the repulsive forces. That is why one must use empirical potentials. The potential derived by the English physicist John Lennard-Jones is the one most widely used.

<https://aiichironakano.github.io/phys516/Battimelli-ComputerMeetsPhysics-Springer20.pdf>, p. 58

cf. Buckingham potential

$$u(r) = A \exp(-Br) - \frac{C}{r^6}$$